

# Effect of Sulfhydryl Modification on Rat Kidney Basolateral Plasma Membrane Transport Function

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**Abstract** Transport processes are the hallmark of functioning kidney. Various nephrotoxics disrupt the transport processes to manifest nephrotoxicity. Of several nephrotoxics, mercuric chloride ( $\text{HgCl}_2$ ) depletes the reduced glutathione (GSH) in kidney and has been observed to affect the *in vitro* *p*-aminohippurate (PAH) transport by basolateral (BL) membrane vesicles. The role of renal nonprotein sulfhydryls such as, reduced GSH has been demonstrated to affect the PAH transport by BL membrane vesicles. The role of protein sulfhydryls in transport process of PAH by BL membrane is not known. Due to mercury mediated effects on sulfhydryls, the effects of protein-sulfhydryls (–SH) modifying reagents in the current study were investigated on PAH transport by BL membrane. It was observed that modification of –SH by *p*-chloromercuribenzoate sulphate (*p*CMBS), and mercuric chloride ( $\text{HgCl}_2$ ) decreased while recovering the protein –SH with dithiothreitol treatment provided protection against the effects of *p*CMBS, and  $\text{HgCl}_2$  on PAH transport by BL membrane vesicles.

**Keywords** Transport · Membrane function · Sulfhydryls · Mercuric chloride

Transport of solutes and organic substances is the hallmark of functioning kidney (Esteve-Font et al. 2012). Many nephrotoxics have been studied over the decades to understand the mechanism of their action (Baggett and Berndt 1986; Klone and Johnson 1983; McDowell et al. 1976; Zalme et al. 1976). Many of these studies have not addressed the cellular and subcellular insults, which results in acute tubular necrosis leading to acute renal failure. One of the major concerns is that how a number of toxicants affecting the tubular epithelial function lead to same sequence of events leading to nephrotoxicity. The failure of functioning kidney after nephrotoxics, suggests a common mechanism beside diverse nature of nephrotoxics. Like the specific role of mitochondria in toxicity and metabolic syndrome, plasma membrane vesicles in the form of functioning transport processes can also provide the opportunity as the subcellular target of nephrotoxics (Houston 2011; Lenaz 2012; Scatena 2012). The availability of enriched preparation of basolateral (BL) plasma membrane vesicles from renal cortical tissues has made it possible to evaluate the role of sulfhydryls on specific transport process (Boumendi-Podevin and Podevin 1983; Jensen and Berndt 1988). A wide variety of toxicants especially nephrotoxics cause toxicity by oxidative stress which can alter both the non-protein and protein –SH status of cells that can contribute to altered function (Lash 2011). Effects on transport processes due to nephrotoxics in acute renal failure have been observed (Hori et al. 1985; Lee et al. 1990). In the present study, the protein –SH level was reduced by treatment with *p*CMBS, and  $\text{HgCl}_2$  and their effects on PAH transport by BL membrane

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vesicles isolated from kidney cortex were studied. Further effects of reducing agent, dithiothreitol (DTT) in counteracting the effects of *p*CMBS, and HgCl<sub>2</sub> on PAH transport by BL membrane vesicles were also studied. It was observed that effects of reduction of membrane associated –SH caused reduction in transport of PAH by BL membrane vesicles while treatment with DTT was able to reverse the effects offering protection against the –SH modifying agents including HgCl<sub>2</sub>.

## Materials and Methods

The right side out BL membrane vesicles were prepared by modification of the method of Boumendil-Podevin and Podevin (Boumendil-Podevin and Podevin 1983). The protein content of the plasma membrane was determined by dye binding method of Bradford (Bradford, 1976). The membrane preparation was diluted to 2.5 mg protein/mL with buffered sucrose solution. Marker enzymes were used to monitor the relative purity of the BL membranes. Na<sup>+</sup>/K<sup>+</sup> ATPase activity was enriched fivefolds–ninefolds in BL membranes relative to kidney cortex homogenate.

The BL membrane vesicles and sheets (after lysing membrane vesicles into water as mentioned in transport measurement) were treated with *p*CMBS, and HgCl<sub>2</sub> for 10 min and the –SH level was measured by Ellman's reagent, 5,5'-Dithio-bis (2-nitrobenzoic acid) (DTNB). A standard of reduced glutathione (GSH) was used to determine the sulfhydryl values expressed as reduced GSH (Baker et al. 1990).

Transport of [<sup>14</sup>C] *p*-aminohippurate ([<sup>14</sup>C] PAH) into BL membrane vesicles was assayed at 37°C into a total volume of 60 µL. BL membrane vesicles were preloaded with PAH (200 µM) after 30 min preincubation with PAH and repelleting the membranes in ultracentrifuge as described earlier followed by resuspension in sucrose-Tris buffer solution (Ansari et al. 1990). BL membrane vesicles of 50 µg protein were added to each vial in 20 µL aliquots, followed by buffered sucrose (270 mM sucrose–20 mM Tris–HEPES). Transport was initiated by addition of labeled [<sup>14</sup>C] PAH. Transport was abrogated after 15, 30, 60, and 600 s by filtering 50 µL aliquots through 0.45 µm filters. The transport was stopped by washing the filters with 6.0 mL ice-cold solution containing 130 mM NaCl, 20 mM Tris containing 5 mM probenecid.

The filters were dried and dissolved in 5.0 mL scintillation fluid. The corrections for the amount of the radioactivity bound to membrane sheets and filters were done by incubating 20 µL of membrane protein in 0.36 mL of water for 5 min, followed by additional 5-minute incubation after

addition of 20 µL labeled substrate. After filtration as mentioned above, the radioactive binding to filter and membrane was determined by counting the filters. Since the net transport values varied from preparation to preparation, the results are expressed as the percentage of the equilibrium, which was achieved and determined at 600 s. To ascertain that equilibrium was achieved at 600 s, the preliminary experiment measurements were performed at 20 and 30 min. The values of 20 and 30 min did not differ from 600 s measurements.

The effects of *p*CMBS, and HgCl<sub>2</sub> on PAH transport by BL membrane vesicles were determined by exposing the BL membranes with *p*CMBS, and HgCl<sub>2</sub> for 10 min. For determining the effects of DTT on transport process of PAH by BL membrane vesicles, initially, membranes were treated with *p*CMBS, and HgCl<sub>2</sub> for 10 min. These agents were removed after centrifugation of the membrane. DTT was added to membrane vesicle for 30 min. After removal of DTT by centrifugation, the transport studies were performed for set period of times as mentioned in graphs.

The data are expressed as mean ± standard error of five or six membrane preparations. The data were subjected to one-way analysis of variance (ANOVA). The individual differences among means were determined by student's *t*-test.

## Results and Discussion

The effects of *p*CMBS and HgCl<sub>2</sub> treatments on sulfhydryl levels of BL membrane vesicles and sheets are presented in Table 1. The –SH modifying agents, *p*CMBS and HgCl<sub>2</sub> were able to decrease the –SH status of BL membrane vesicles and sheets. Both *p*CMBS and HgCl<sub>2</sub> treatment resulted in statistically significant reduction of –SH levels of BL membranes and sheets. The nephrotoxicant, HgCl<sub>2</sub> was able to deplete significantly the –SH level of basolateral membrane vesicles and sheets like other –SH modifying reagent, *p*CMBS.

**Table 1** SH status of BL membrane vesicles and sheets of control and treatment groups

Group	Membrane vesicles	Sheets
Control	38.03 ± 6.49	40.38 ± 6.67
10 µM <i>p</i> CMBS	25.19 ± 3.00*	22.25 ± 9.12*
100 µM <i>p</i> CMBS	15.63 ± 4.26**	6.99 ± 7.97**
Control	34.74 ± 4.69	37.14 ± 4.88
10 µM HgCl <sub>2</sub>	23.03 ± 5.96*	13.62 ± 5.51*
100 µM HgCl <sub>2</sub>	15.07 ± 7.12**	8.48 ± 2.16**

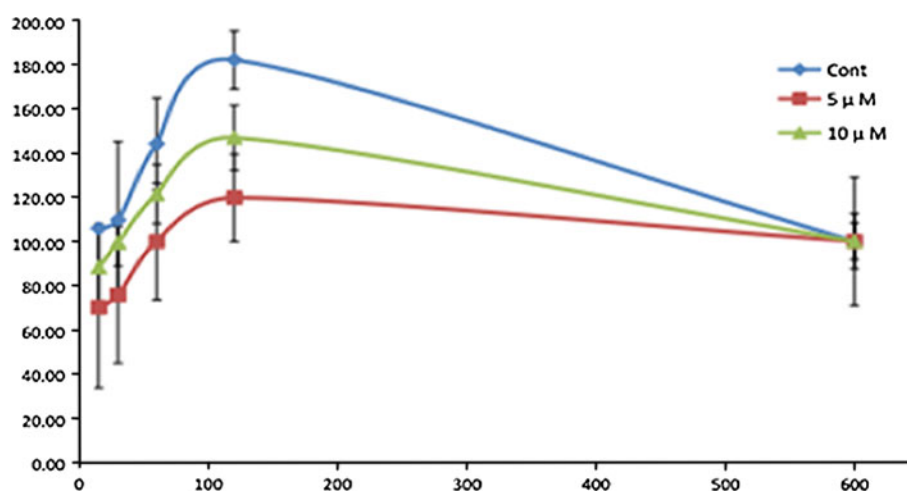
Data are presented as mean (nmoles GSH/mg protein) ± SD of 5 observations. \* *p* < 0.001 and \*\* *p* < 0.0001

The effects of *p*CMBS, and  $\text{HgCl}_2$  on PAH transport by BL membrane vesicles are presented in Figs. 1, 2 and 3. Initial treatment of membrane vesicles with *p*CMBS at 5 and 10  $\mu\text{M}$  did not significantly reduce the overshoot value of transport (60 and 120 s transport values) (Fig. 1). However, later treatment of BL membrane vesicles with *p*CMBS and  $\text{HgCl}_2$  at 10  $\mu\text{M}$ , significantly reduced the percent transport values at 60 and 120 s. The protective effects of DTT treatment on *p*CMBS, and  $\text{HgCl}_2$  mediated effects on PAH transport by BL membrane are summarized in Figs. 2 and 3 respectively. The DTT treatment of vesicles for 30 min after 10 min pretreatment of vesicles with *p*CMBS, and  $\text{HgCl}_2$  offered protection to PAH transport by BL membrane. DTT was able to reverse the effects of *p*CMBS, and  $\text{HgCl}_2$ . It is noteworthy that DTT not only restored the overshoot at 2 min of PAH transport but also increased the overshoot level from control.

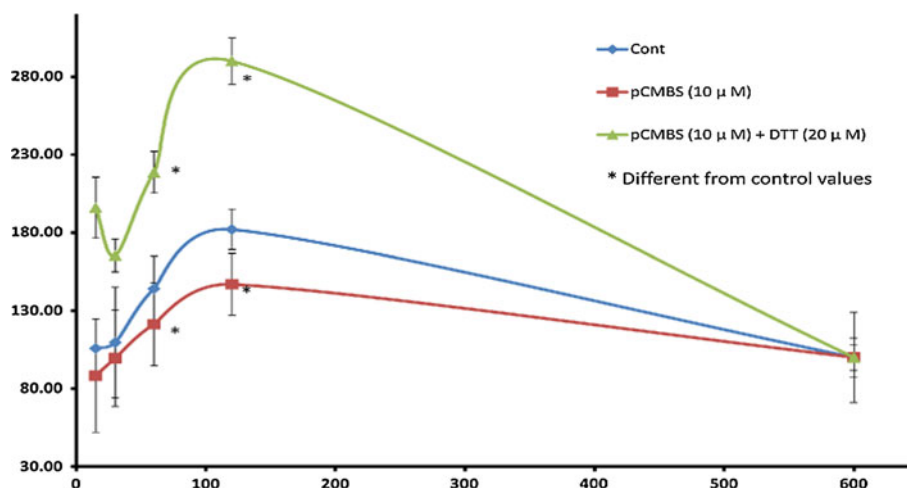
Isolation of membrane vesicles has provided the opportunity to study the effects of nephrotoxics on membrane functions especially on the transport process. It

also, provides the opportunity to study the effects of toxicants on transport processes devoid of metabolic effects. Often the toxicants affect the metabolic process that disrupts the energy process involved in active transport processes. Earlier, our studies have demonstrated that nephrotoxics disrupt the transport processes of both BL and brush border (BB) membrane vesicles (Ansari et al. 1990). One of the nephrotoxics,  $\text{HgCl}_2$  is capable of binding with proteins –SH and C=O groups and affects the transport of PAH by BL membrane vesicles when studying in vitro exposure and transport studies (Ansari et al. 1990). However, when animals were exposed with  $\text{HgCl}_2$ , it affected  $\text{Na}^+$ /glucose transport by BB membrane vesicle after 16 h of exposure (Ansari et al. 1990). In order to evaluate the mechanism of effects of  $\text{HgCl}_2$ , we utilized –SH modifying agents to evaluate the effects on PAH transport process by BL membrane vesicle. It was observed that both *p*CMBS and  $\text{HgCl}_2$  significantly reduced the amount of –SH which is expressed as reduced GSH level (Table 1). Treatment with *p*CMBS, and  $\text{HgCl}_2$  also

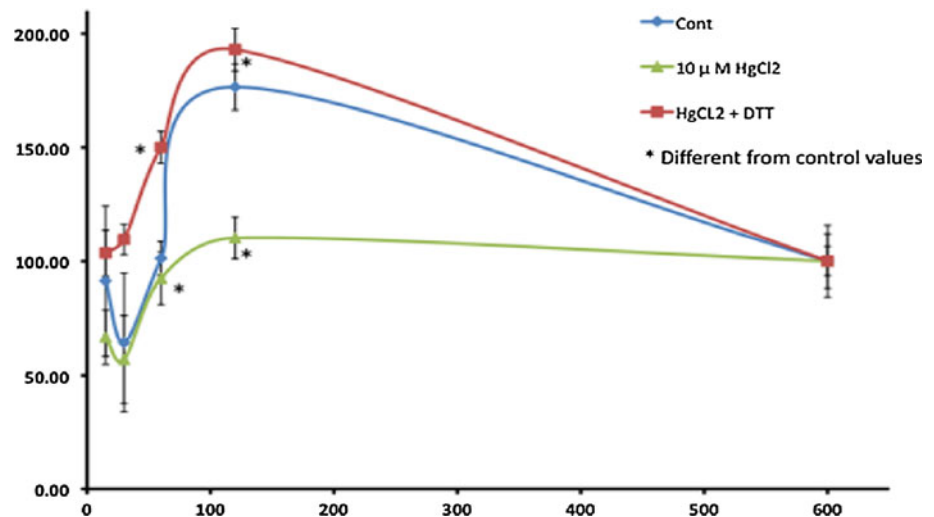
**Fig. 1** Effect of *p*CMBS on PAH Transport by BL membrane vesicles



**Fig. 2** Effect of DTT (20  $\mu\text{M}$ ) on *p*CMBS (10  $\mu\text{M}$ ) induced effect on PAH transport by BL membrane vesicles



**Fig. 3** Effect of DTT (20  $\mu$ M) on  $\text{HgCl}_2$  (5–10  $\mu$ M) induced effect on PAH transport by BL membrane vesicles



decreased the overshoot value of the PAH transport by BL membrane vesicles (Figs. 1, 2, 3). Impairment of cellular redox status has been found to affect the transport of PAH by brush border membrane vesicles in ischemic acute renal failure (Montagna et al. 1998). In an attempt to recover the  $-\text{SH}$  groups of the proteins of membranes,  $p\text{CMBS}$ , and  $\text{HgCl}_2$  (10 min) treated membranes were treated with DTT for 30 min and transport studies of PAH with BL membranes demonstrated the recovery of the transport process (Figs. 2, 3). The treatment of membrane vesicles with DTT, which were pretreated with  $p\text{CMBS}$  and  $\text{HgCl}_2$ , not only restored the transport overshoot values to control level but also further enhanced the values of overshoot over the control. The enhancement of overshoot values by DTT over the control values indicates that certain  $-\text{SH}$  groups may have been oxidized in membrane proteins during preparation which may have caused effects on transport process. Earlier, reduction in nonprotein sulfhydryls, i.e., GSH by diethylmaleate and buthionine sulfoximine (BSO) and enhancement of level of GSH by glutathione monomethyl ester (GMEE) has been shown to affect the transport of PAH by BL membrane vesicles. The reduction of non-protein sulfhydryls has shifted the overshoot of transport of PAH from 120 to 60 s. (Ansari et al. 1991). A detailed analysis of effects of DTT on membrane isolation and its effects on transport is required.

The transport of PAH occurs via organic anion transport system (Riedmaier et al. 2012). The human PAH transporter has long been cloned (Hosoyamada et al. 1999; Reid et al. 1998). Organic anion transporters from rat and other species have also been cloned and their function defined (Sekine et al. 1997; Sweet et al. 1997). Human organic anion transporter (hOAT1) has been found sensitive to sulfhydryls modifying agents (Astorga et al. 2011). A specific cysteine (C440) mutation in between hOAT transporter is important for  $\text{HgCl}_2$  sensitivity (Astorga et al.

2011). Similarly, mouse organic anion transporter 1 (mOAT1) has also been found sensitive to cysteine modification by  $p\text{CMBS}$  (Tanaka et al. 2004). Earlier, sulfhydryl reagents have been demonstrated to affect the PAH transport (Dantzler and Bentley 1983; Tse et al. 1983). The restoration of transport process by treatment with sulfhydryl reducing agents, like DTT after sulfhydryl modification with  $p\text{CMBS}$ , and  $\text{HgCl}_2$  has not been demonstrated. In an earlier study, we demonstrated that effects of  $\text{HgCl}_2$  on brush border mediated  $\text{Na}^+$ /glucose transport is observed by 16 h while recovery occurs by 48 h. This recovery is possible, probably, with the restoration of non-protein sulfhydryls especially GSH, which may provide protection against sulfhydryl modification of membrane transporters (Ansari et al. 1990). The effects of these  $-\text{SH}$  modifying agents indicate that nephrotoxicant, such as  $\text{HgCl}_2$  and possibly other toxicants producing oxidative stress in kidney epithelial cells may be acting after modifying the  $-\text{SH}$  groups of transporters and compromising the transport function of membrane. The study demonstrates that kidney PAH epithelial transport process can recover after restoration of sulfhydryl status of membrane transport systems which could be target of nephrotoxicants especially  $\text{HgCl}_2$  and other xenobiotics which exhibit their toxicities by producing oxidative stress.

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